# 502 35100

量測原理與機工實驗

IV. Laser Diagnostics for Thermofluid Measurements

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- Background and principle
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- A study of measuring flow velocity and species

- Measurements of temperature
- and species concentration
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- Raman scattering
- CARS
- LIFS

### References :

R. J. Goldstein, Fluid Mechanics Measurements, Hemisphere, 1983 S. Tavoularis, Measurement in Fluid Mechanics, Cambridge Univ., 2005 A. C. Eckbreth, Laser Diagnostics for Combustion Temperature and Species, Abacus Press, 1988

## Introduction to Laser

- Acronym: light amplification by stimulated emission of radiation
- Advantages as compared with thermal light sources
  - Coherent (with all light fronts in phase)
  - Collimated and concentrated (essentially parallel, with a small cross-sectional area)
  - Monochromatic (with spectral energy concentrated in one or more extremely narrow bands)
- Disadvantages
  - Diffraction due to dust particles → *speckles*: distortion by fringes and patterns; partly removed by a pinhole (spatial filter)

### Principle

- Active medium: gas, crystal, semiconductor, liquid solution ...
- Medium composed of particles (atoms, ions, or molecules), with electrons existing only at specific quantized energy levels
- absorption and spontaneous emission: random direction
- *stimulated emission*: coherent emission of an incident photon,
   with identical frequency (energy), phase, and direction
- *population inversion*: for effective stimulated emission, larger population of atoms at a high energy state than at a lower state
- Use (nonintrusive): flow visualization, measurements for velocity, pressure, temperature, and composition, both in liquids and gases









- Laser types (Cont'd)
  - *Helium-neon* (He-Ne) lasers: CW type, power 0.3 15 mW, at  $\lambda = 633$  nm (red)
  - easy operation and low cost, flow visualization
  - active medium composed of helium and neon atoms
  - energy pump source of high-voltage (~ 2000 V) electric field
  - Argon-ion (Ar) lasers: CW type, power 100 mW 10 W, at 7
  - wavelengths particularly  $\lambda$  = 488 (blue) and 514.5 nm (green)
  - low power efficiency, water/air cooling
  - for flow visualization, LDV, PDPA, PIV ...
  - active medium of argon atoms, at ion state with collision by electrons accelerated via *E* field, in a plasma tube





• *CO*<sub>2</sub> lasers:

– active medium of CO<sub>2</sub> molecules at ground state

- oscillate at  $\lambda$  = 10.4 µm (infrared), either in continuous or pulsed mode - population inversion to higher vibrational energy states caused by

collisions with nitrogen molecules at a metastable, excited vibrational mode; pumped thermally

• Laser diodes:

 semiconductor devices that emit coherent radiation in visible or infrared ranges when current passes through

– used extensively in optical-fiber communication systems, compact disc players, laser printers, remote controls, and intrusion detection systems

- much smaller than conventional lasers and lower power requirement

- high-power laser diodes used for flow visualization, LDV, PIV ...



# LDV for Velocity Measurement Acronym: laser Doppler velocimetry (LDV) or anemometry (LDA) Advantages as compared with conventional techniques Non-intrusive, linear relation between flow velocity and Doppler frequency (no calibration and errors reduced), rapid response and high temporal resolution, high spatial resolution (1 mm<sup>3</sup>), large speed range, reversed flow, temperature insensitive Disadvantages Transparent windows, seeding particles that follow flows (rate), complex alignment, not-so-easy signal processing with weak scattering light and interruption from different particles (low SNR)

























- Mie scattering
  - Elastic scattering of light quanta from particulate matter
  - Basic effect underlying particle sizing, LDV, differential absorption backscattering ~ local concentration of particles

• Accuracy affected by particle size (polydispersity), refractive index, particle coagulation, and absorption; not specific to species

- Molecular radiation emission
  - Radiation scattering at the molecular level
  - Identification of species and concentration
  - $I_{nm} = A_{nm}N_nhv_{nm}$ , where  $A_{nm}$  is Einstein transition probability
  - High-power radiation sources and careful experimental setting

### Rayleigh scattering

- Elastic scattering from isolated atoms and ordinary molecules:  $D/\lambda \ll 1 \ (< 1/15) \sim \text{limit of Mie scattering}$
- Sky colors  $\leftarrow$  Rayleigh scattering cross section  $\sigma_R = \sigma_T (\lambda_0 / \lambda)^4$
- where *Thomson-scattering cross section* for a single electron  $\sigma_T$  = 6.65×10<sup>-29</sup> m<sup>2</sup>, and  $\lambda_0$  the characteristic wavelength of an atom
- Interfered by Mie scattering ~ clean condition, strong signal ...
- Not specific to species; continuous signal, temporal resolution
- Measurement for total density, temperature (const. *P*, spectra)
- Different from Mie scattering in which many atoms/molecules gather and EM waves interfere (no lateral scattering, directional)





- Spontaneous Raman (SRS) effect (~ 10<sup>-14</sup> 10<sup>-12</sup> s)
- Measurement of temperature, species and concentration

- Unique vibrational energy levels, not contaminated by incident light,

- detection for different molecules; low intensity  $(10^{-5} 10^{-2} \text{ lower than RS})$
- As  $T^{\uparrow}$ , vibrational/rotational energy bands  $\uparrow \rightarrow$  spectral overlapping
- High-power (e.g. KrF excimer) laser; low  $\lambda$  preferred,  $\sigma \propto \lambda^4$
- Near-resonant Raman scattering (cross section enhanced by 6 orders)
- Frequency tuned near (off) an electronic resonance of molecule, nearly instantaneous, insensitive to collisional quenching
- Stokes transition: photon emitted at lower energy than absorbed one
- Anti-Stokes transition: photon emitted at higher energy than absorbed







### Fluorescence

• *Fluorescence*: time delay after absorption ~  $10^{-10} - 10^{-5}$  s; transition between the same electronic spin states (multiplicity)

• *Phosphorescence*: time delay ~  $10^{-4}$  s – hours; transition between different electronic spin states

• An uncertainty – *quenching* in fluorescence: with sufficient time for collisions of molecules to occur and photon energy to be converted to chemical reaction, dissociation, and ionization energies, before emission

- Other causes to fluorescence in addition to photon absorption: electron bombardment, heating/chemical reaction (*chemiluminescence*)
- Resonant/non-resonant(shifted) fluorescence: latter preferred, to allow isolation of fluorescence radiation from incident light & Mie scattering

• High intensity, independent of direction, randomly polarized

- Laser-induced fluorescence spectroscopy (LIF/LIFS)
- Principal measurement of concentration in liquid and gas
- For *T*: fluorescence quantum efficiency  $\downarrow$  with *T* (~linearly)
- Type: local LIF vs. planar LIF (PLIF)
- Qualitative and quantitative (limited by depletion, extinction, *T*
- sensitivity, pH sensitivity, spatial resolution, non-uniform illumination)
- Use passive fluorescent seeding materials in non-reacting gases such as biacetyl which emits radiation in visible range
- Mostly used in flames, with fluorescence given by radicals such as OH, CH, CO, and NO  $\rightarrow$  map chemical reactions
- Produce strongest radiation signal from a single species
- Excited by tunable, pulsed, dye and excimer lasers, visible/ultraviolet

# A Case Study of Measuring Velocity and Species Concentration

- A study of bluff-body combustion field
  - Background
  - Experimental facility
- Measurements
  - Visual observation
  - Flow velocity field by LDV
  - Distribution of OH radicals by PLIF













